

Electronic Supplementary Material

Photocatalytic syngas synthesis from CO₂ and H₂O using ultrafine CeO₂-decorated layered double hydroxide nanosheets under visible-light up to 600 nm

Ling Tan*, Kipkorir Peter*, Jing Ren, Baoyang Du, Xiaojie Hao, Yufei Zhao (✉), Yu-Fei Song (✉)

State Key Laboratory of Chemical Resource Engineering, Beijing University of Chemical Technology, Beijing 100029, China

E-mails: zhaoyufei@mail.buct.edu.cn (Zhao Y); songyf@mail.buct.edu.cn (Song Y)

Table S1. BET surface areas of LDH, Ce-0.15, Ce-0.20 and Ce-0.40

sample	BET surface area/(m ² ·g ⁻¹)
LDH	108.84
Ce-0.15	116.88
Ce-0.20	110.74
Ce-0.40	58.92

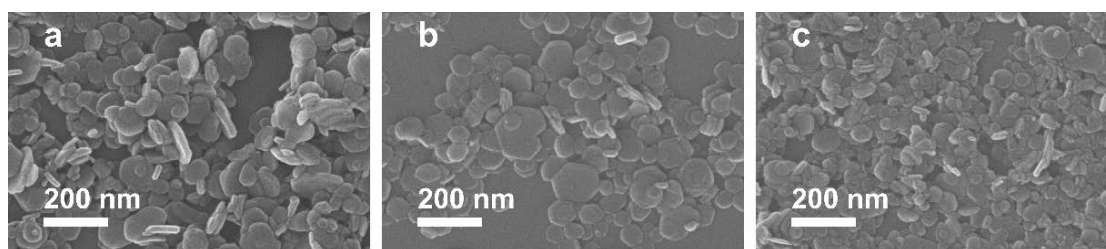


Figure S1. SEM images of (a) LDH (b) Ce-0.15 and (c) Ce-0.40.

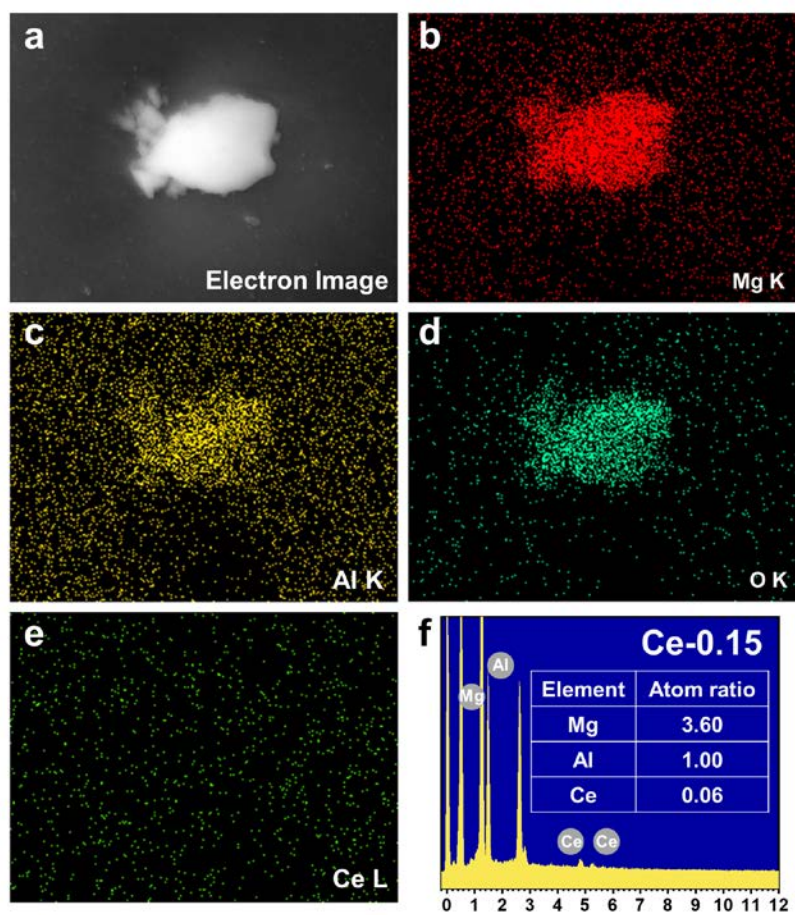


Figure S2. (a) SEM image of Ce-0.15; Energy-dispersive X-ray (EDS) elemental mapping of Ce-0.15 for (b) Mg, (c) Al, (d) O and (e) Ce, respectively; (f) EDS spectroscopy of Ce-0.15.

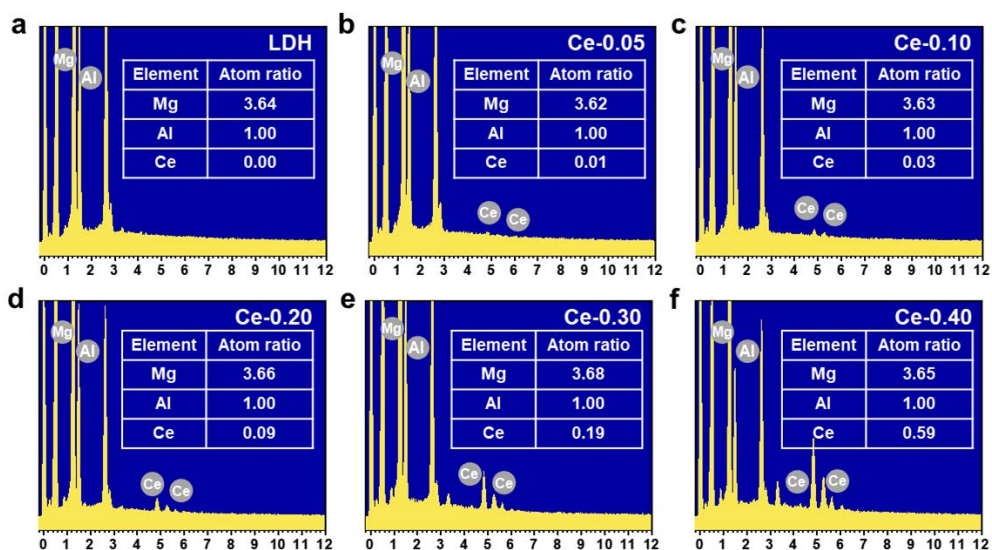


Figure S3. Energy-dispersive X-ray spectroscopy of (a) LDH, (b) Ce-0.05, (c) Ce-0.10, (d) Ce-0.20, (e) Ce-0.30 and (f) Ce-0.40, respectively.

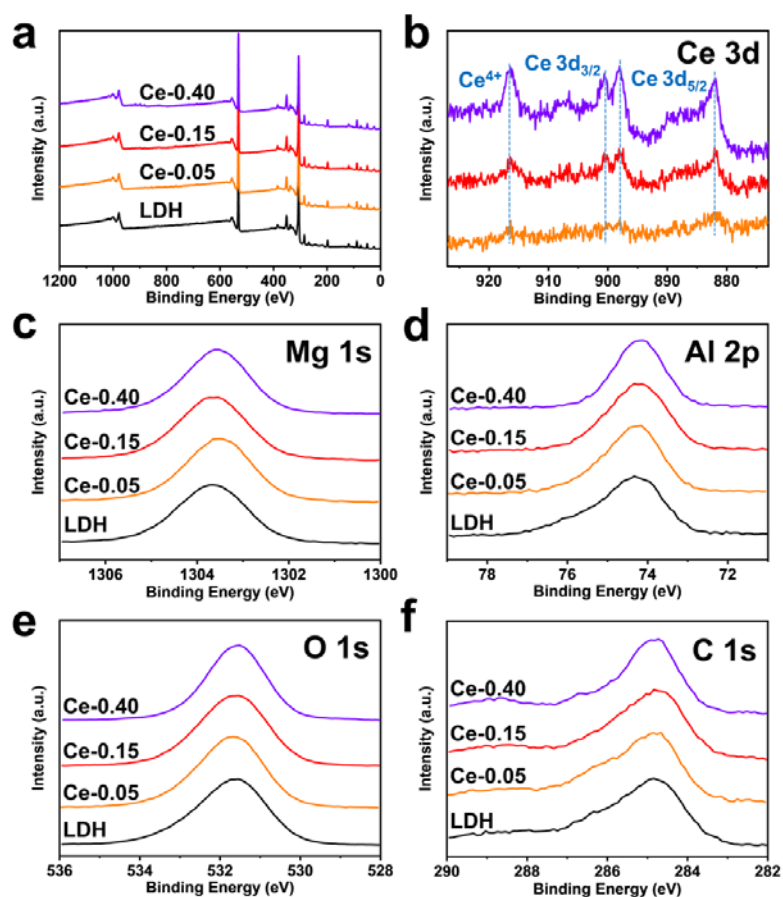


Figure S4. XPS (a) survey spectra and (b) Ce 3d spectra for LDH, Ce-0.05, Ce-0.15 and Ce-0.40.

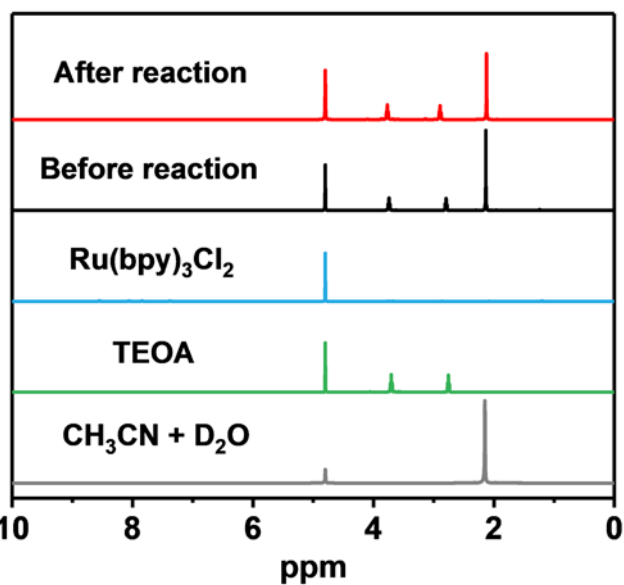


Figure S5. $^1\text{H-NMR}$ spectra of CH_3CN , TEOA, $\text{Ru}(\text{bpy})_3\text{Cl}_2 \cdot 6\text{H}_2\text{O}$ and liquid sample, taken from the reaction system (irradiating with $\lambda > 400$ nm for 1 h).

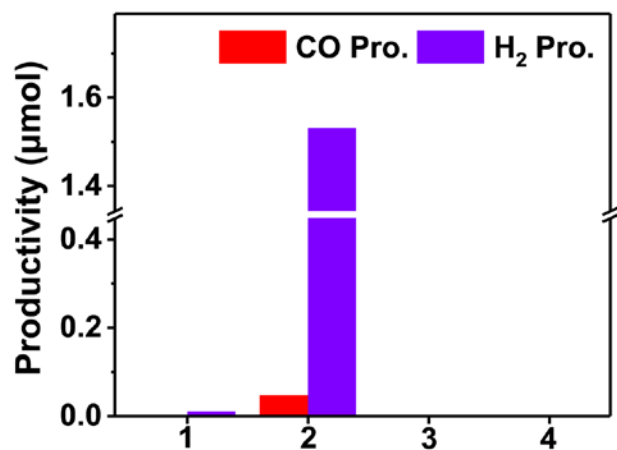


Figure S6. The productivity of catalyst under control reaction conditions experiments, note: 1. Ar atmosphere, 2. Without Ce-0.15, 3. Without Ru(bpy)₃Cl₂, 4. In dark.

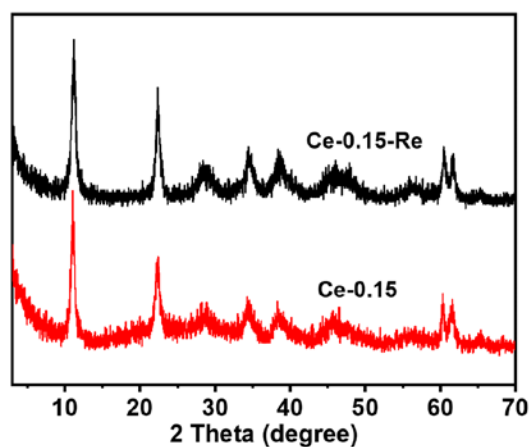


Figure S7. XRD patterns of Ce-0.15 and the recycled Ce-0.15 (denoted as Ce-0.15-Re).

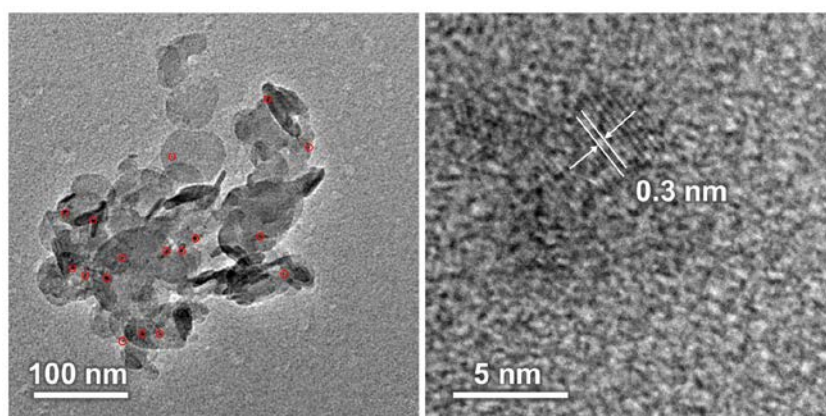


Figure S8. (a) TEM image and (b) HRTEM image of recycled Ce-0.15.

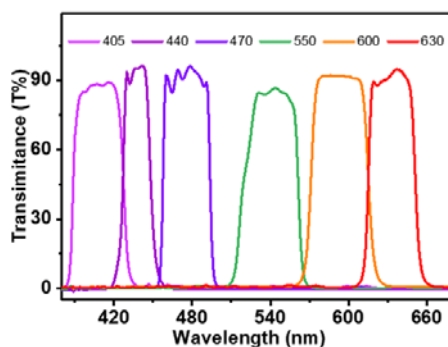


Figure S9. The monochromatic light at different wavelength adapted from our previous work.[1]

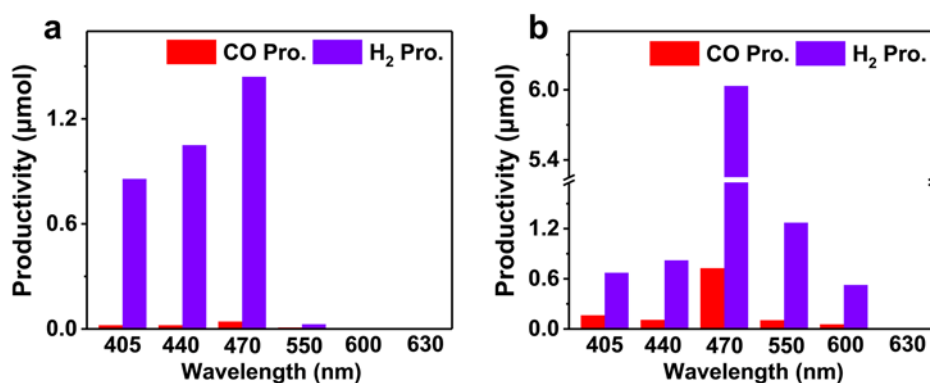


Figure S10. The productivity of (a) LDH and (b) Ce-0.15 in CO₂PR under different cut-off filter light irradiation.

The External Quantum Efficiency (EQE) values of CO was calculated using the equation as reported:[2]

$$EQE(\%) = \frac{2n \times N_A \times h \times c}{t_{irr} \times \lambda \times I \times A} \times 100$$

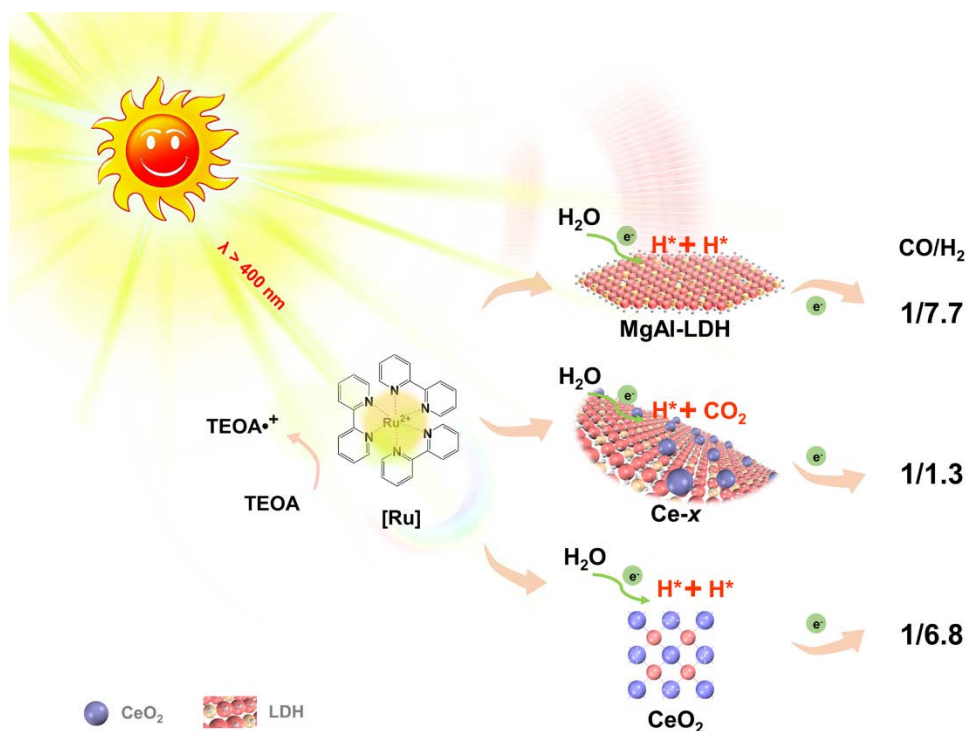
Where n is the amount of CO produced per time, N_A is Avogadro's constant, h is the Planck constant, c is the speed of light, t_{irr} is the irradiation time, λ is the irradiation wavelength, I is the irradiation intensity and A is the irradiated area. So, the EQE values of CO under different wavelength irradiation was shown as Table S2.

Table S2. The External Quantum Efficiency (EQE) values of CO under different wavelength irradiation by LDH and Ce-0.15

Sample	405 nm	470 nm	550 nm	600 nm
LDH (%)	0.05	0.04	0.07	0
Ce-0.15 (%)	0.43	1.45	0.16	0.05

Table S3. Comparison of photocatalytic CO₂ reduction performance for various photocatalysts in this work and in previous literature

	photocatalyst	photosensitizer co-catalyst	sacrificial agent	solvent	light source	evolution rate of major product ($\mu\text{mol g}^{-1} \text{h}^{-1}$)	reference
1	Ce-0.15	Ru(bpy) ₃ Cl ₂ ·6H ₂ O	TEOA	MeCN-H ₂ O (3:1 v/v)	300 W Xe ($\lambda = 600 \text{ nm}$)	CO: 5 (8.7%) H₂: 52 (91.3%)	This work
2	Pd/CoAl-7.57	Ru(bpy) ₃ Cl ₂ ·6H ₂ O	TEOA	MeCN-H ₂ O (3:1 v/v)	300 W Xe ($\lambda > 600 \text{ nm}$)	CO: 4.1 (43.2%) H ₂ : 5.4 (56.8%)	JEC, 2020, 46, 1[3]
3	CoAl-LDH	Ru(bpy) ₃ Cl ₂ ·6H ₂ O	TEOA	MeCN-H ₂ O (3:1 v/v)	300 W Xe ($\lambda = 600 \text{ nm}$)	CO: 43730 (55.8%) H ₂ : (44.2%)	IECR, 2020, 59, 5848[4]
4	ZnIn ₂ S ₄ -In ₂ O ₃	CoCl ₂ 2'2-bipyridine	TEOA	MeCN-H ₂ O (3:2 v/v)	300 W Xe ($\lambda > 400 \text{ nm}$)	CO: 3075 (~75%) H ₂ : ~800 (~25%)	JACS, 2018, 140, 5037[5]
5	Co-G nanosheet	Ru(bpy) ₃ Cl ₂ ·6H ₂ O	TEOA	MeCN-H ₂ O (3:1 v/v)	300 W Xe ($\lambda > 420 \text{ nm}$)	CO: (79.4%) H ₂ : (20.6%)	Adv. Mater. 2018, 1704624[6]
6	Ni MOLs (Ni MOF monolayers)	Ru(bpy) ₃ Cl ₂ ·6H ₂ O	TEOA	MeCN-H ₂ O (3:2 v/v)	5 W LED ($\lambda > 400 \text{ nm}$)	CO: 12500 (97.8%) H ₂ : 280 (2.2%)	Angew, 2018, 57,16811[7]
7	MAF-X27I-OH	Ru(bpy) ₃ Cl ₂ ·6H ₂ O	TEOA	MeCN-H ₂ O (4:1 v/v)	LED ($\lambda > 420 \text{ nm}$)	CO: 45 μmol (98.3%) H ₂ : 0.8 μmol (1.7%)	JACS, 2018, 140, 38[8]
8	Dye/TiO ₂ /Re(I): Co(III) complex (9:1)	dye	BIH	DMF-H ₂ O (4:1 v/v)	60 W LED ($\lambda > 400 \text{ nm}$)	CO: (76.9%) H ₂ : (23.1%)	Angew, 2017, 56, 976[9]
9	In ₂ S ₃ -CdIn ₂ S ₄	Co(bpy) ₃ ²⁺	TEOA	MeCN-H ₂ O (3:2 v/v)	300 W Xe ($\lambda > 400 \text{ nm}$)	CO: 825 (70.2%) H ₂ :~350 (29.8%)	JACS, 2017, 139, 17305[10]
10	Co ₃ O ₄ -400	Ru(bpy) ₃ Cl ₂ ·6H ₂ O	TEOA	MeCN-H ₂ O (3:1 v/v)	300 W Xe ($\lambda > 420 \text{ nm}$)	CO: 2003 (77.1%) H ₂ : 595 (22.9%)	Adv. Mater. 2016, 28, 6485[11]
11	C doped BN	CoCl ₂ 2'2-bipyridine	TEOA	MeCN-H ₂ O (2:1 v/v)	300 W Xe ($\lambda > 420 \text{ nm}$)	CO: 9.3 (76.2%) H ₂ : 2.9 (23.8%)	Nature Commun., 2015, 6, 7698[12]



Scheme S1. The mechanism schematic of the photocatalytic CO₂ on LDH, Ce-*x* and CeO₂, respectively.

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