

Electronic Supplementary Material

Improving hole transfer of boron nitride quantum dots

modified PDI for efficient photodegradation

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1. Experimental Section

1.1 Characterization

The crystal structure of the samples was performed by X-ray diffraction (XRD) with Cu $K\alpha$ radiation with $\lambda = 0.15418$ nm (Bruker D8 Advance). The Fourier transform infrared spectroscopy (FTIR) spectra were obtained by a Bruker spectrometer. The X-ray photoelectron spectroscopy (XPS) measurements were obtained from an ESCALAB 250 Xi instrument (Thermo Scientific) with Al $K\alpha$ radiation under the light irradiation at $\lambda = 365$ nm (the C 1s = 284.60 eV binding energy is used as the energy standard for charging correction). The high-resolution transmission electron microscope (HR-TEM) (JEOL JEM-F200), scanning electron microscopy (SEM) (Hitachi S-4800), and were used to perform morphology and microstructure. Atomic Resolution analytical Microscope (AFM) was carried out

using Cypher VRS with Kelvin probe (HQ NSC18/Pt). The photoluminescence spectra (PL) spectra were recorded by a fluorescence spectrophotometer (Hitachi F-4600) made in Japan. The time-resolved fluorescence decay spectra (TRFL) were obtained on a time-resolved Fluorescence Spectrometers (FLS1000, Edinburgh Instruments). The UV-vis diffuse reflectance spectra (DRS) were acquired from the Hitachi U-3900 with BaSO₄ as the reflectance standard.

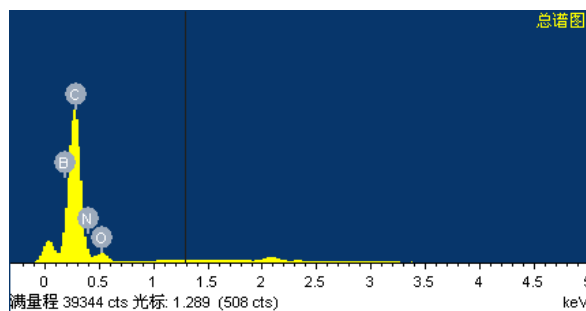


Figure. S1 Elemental distributions of PDI/BNQDs-10.

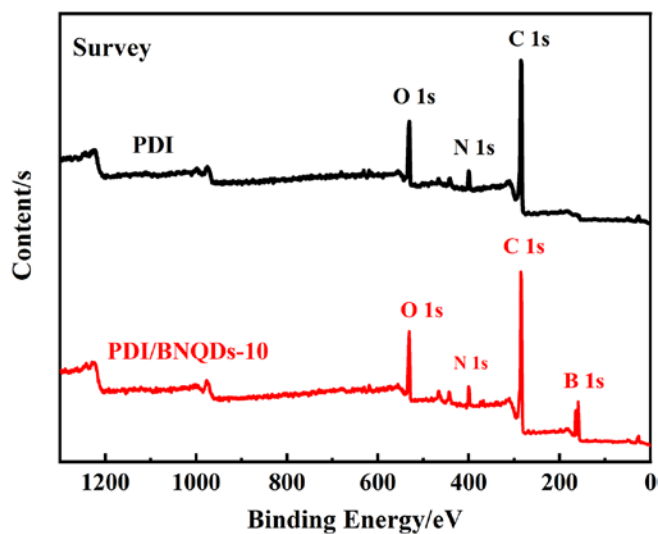


Figure. S2 XPS broad spectra of PDI supramolecular and PDI/BNQDs-10.

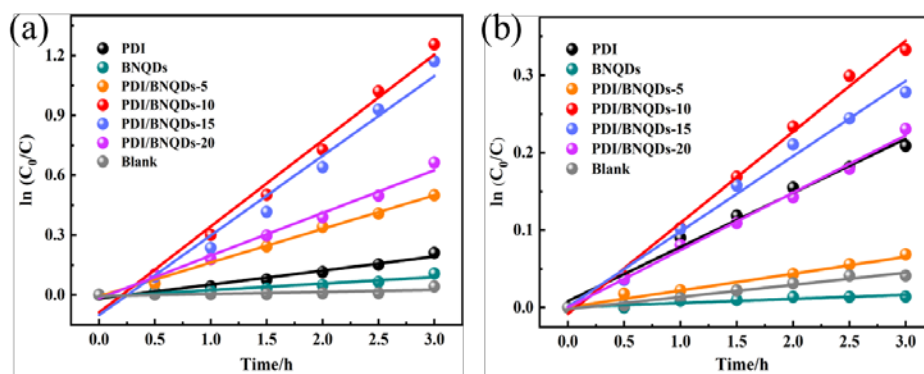


Figure. S3 The corresponding first order kinetics curve fitting of samples under visible light ($\lambda > 420$ nm) (a) MB and (b) CI.

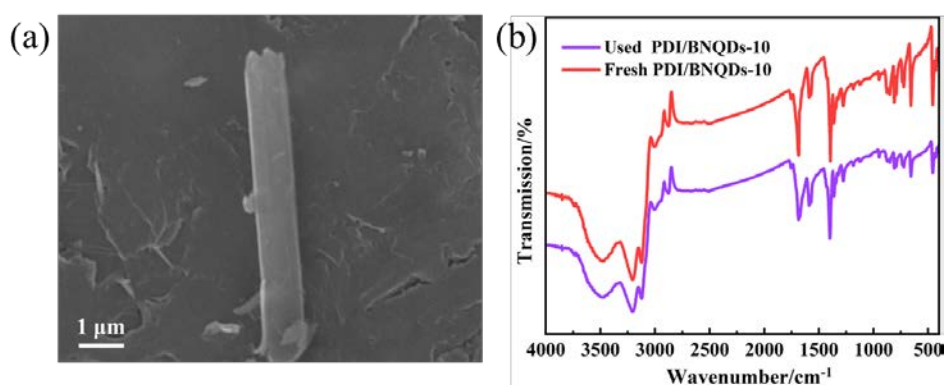


Figure. S4 (a) SEM image, (b) FT-IR spectra of PDI/BNQDs-10 before and after degradation reaction.

Table S1 Chemical composition of PDI/BNQDs-10.

| Percentage of element atoms | | | |
|-----------------------------|-------|------|------|
| C | B | N | O |
| 64.14 | 21.62 | 9.76 | 4.48 |

Table S2 Z potential of PDI supramolecular, BNQDs, and PDI/BNQDs-10.

| Sample | Z potential-1 | Z potential-1 | Z potential-3 | Average potential |
|--------------|---------------|---------------|---------------|-------------------|
| PDI | -2.89 mV | -2.59 mV | -3.45 mV | -2.98 mV |
| BNQDs | -7.10 mV | -7.01 mV | -9.72 mV | -7.94 mV |
| PDI/BNQDs-10 | -60.5 mV | -59.2 mV | -60.8 mV | -60.2 mV |

Table S3 Comparison of the photocatalytic performance of MB and CI by different photocatalysts.

| Material | Catalyst dose (mg·L ⁻¹) | MB (mg·L ⁻¹) | Efficiency (MB) | Efficiency (CI) | Ref. |
|---|-------------------------------------|--------------------------|-------------------|-------------------------|-----------|
| PCP2-PDI | 0.25 | 5 | 78% at 200 min | | [1] |
| Co ₄ I ₃ O ₂₄ H ₁₅ ·3H ₂ O | 0.75 | 8 | 73% at 60 min | | [2] |
| ZnIn ₂ S ₄ /In ₂ O ₃ | 0.4 | 50 | 84.5% at 240 min | | [3] |
| Cds@PANI/CFs | 1.0 | 20 | 77.35% at 300 min | | [4] |
| PDI/BNQDs | 0.7 | 10 | 75% at 180 min | K=0.118 h ⁻¹ | This work |

References

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