Electronic Supplementary Material



Fig. S1 Diagrams of hydrogen production performance of prepared samples.
(a) Time course hydrogen evolution of powder WO₃/CdS; (b) comparison of hydrogen production rate of different samples.



Fig. S2 SEM images of WCd-CT after reaction.

	Sacrificial agent	Light source	H ₂ production/	Ref.
Photocatalyst			$(\mu mol {\cdot} h^{-1} {\cdot} g^{-1})$	
CdS/WO ₃ /CT	Na ₂ S+Na ₂ SO ₃	300 W Xe lamp, AM	4320	This work
		1.5		
CdS/Pt/WO ₃	Lactic acid	500 W Xe lamp, $\lambda >$	2900	[1]
		400 nm		
CdS/CdWO ₄	Na ₂ S+Na ₂ SO ₃	300 W Xe lamp, AM	2400	[2]
		1.5		
CdS/ZnO	Na ₂ S+Na ₂ SO ₃	350 W Xe lamp	4134	[3]
WO3@MoS2/Cd	Lactic acid	300 W Xe lamp	8200	[4]
S				
CdS QDs/CeO ₂	Na ₂ S+Na ₂ SO ₃	300 W Xe lamp, $\lambda >$	101	[5]
		300 nm		
$CdS@ZnIn_2S_4\\$	/	300 W Xe lamp, $\lambda >$	540	[6]
		400 nm		
CdS/MnS	Na ₂ S+Na ₂ SO ₃	300 W Xe lamp, $\lambda >$	1595	[7]
		400 nm		
CdS/MoS ₂	Lactic acid	300 W Xe lamp	2107	[8]
QDs/ZnIn ₂ S ₄				

Table S1 Comparison of catalytic activities of different Z-scheme systems

References

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